esters which have all substituents hydrophobic. These steric effects protect the H atom against protonacceptor elements in the reaction mixture. Computation of a least-squares plane through three atoms surrounding the a carbon $[\mathrm{N}(5), \mathrm{C}(10)$ and $\mathrm{C}(7)]$ reveals that above this plane are located the phenyl ring and the guaiacyl ring with the methoxy group above them. The isoleucyl chain is situated entirely on the other side of this plane. The dihedral angle between least-squares planes of the aromatic rings is $89.4(9)^{\circ}$, in contrast to the $1.6^{\circ}$ reported for the $p$-nitrophenyl ester (ZLNP), thus suggesting different spatial locations of the bulky substituents in the two esters.

The insertion of a calculated amide H atom with an $\mathrm{N}-\mathrm{H}$ distance of $0.94 \AA$ leads to the possibility (if the H atom position is approximately correct) of a weak intermolecular hydrogen bridge $\mathrm{N}(5)-\mathrm{H} \cdots \mathrm{O}(4)^{\prime}$ [atom $\mathrm{O}(4)^{\prime}$ is generated by a $b$-axis translation]. The $\mathrm{N}(5)-\mathrm{O}(4)^{\prime}$ distance is $3.024(5) \AA$ and the $\mathrm{H} \cdots \mathrm{O}(4)^{\prime}$ distance would be $2 \cdot 10 \AA$, forming a $D-\mathrm{H} \cdots A$ angle of $166^{\circ}$. Similar parameters for the intermolecular hydrogen bridge were found for $\mathrm{CBz}-\mathrm{Gly}-\mathrm{ProOH}$ (Tanaka et al., 1977), with a $D-A$ distance of $2.906 \AA$, $\mathrm{H} \cdots A=2 \cdot 16 \AA$ and a $D-\mathrm{H} \cdots A$ angle of $167^{\circ}$.

There are few intermolecular distances less than 3.5 $\AA$ in the crystal lattice, of which the shortest are the molecular interactions between the $\mathrm{O}(15)$ atom of the nitro group (asymmetric unit) and the 4 -nitroguaiacyl group of the molecule generated by the symmetry operation $1-x, \frac{1}{2}+y, 1-z$. The distances are: $\mathrm{O}(15)-\mathrm{N}(14)^{\prime} 2 \cdot 905(5), \mathrm{O}(15)-\mathrm{O}(15)^{\prime} 3 \cdot 163$ (5), and
$\mathrm{O}(15)-\mathrm{C}(34)^{\prime} 3.271$ (6) $\AA$. This may be caused by a polar resonance effect of the nitro group.

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# $D: C$-friedo- $B^{\prime}: A^{\prime}$-neo-Gammacer-9(11)-ene* 

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#### Abstract

C}_{30} \mathrm{H}_{50}\), orthorhombic, $P 2_{1} 2_{1} 2_{1}, a=$ 7.669 (6) , $b=10.571(7), c=32.01(5) \AA(\lambda=$ $1.5418 \AA, T=293 \mathrm{~K}), V=2639.5 \AA^{3}, Z=4, M_{r}=$ $410.4, D_{x}=1.03 \mathrm{Mg} \mathrm{m}^{-3}, F(000)=920, \mu\left(\mathrm{Cu} K_{\mathrm{ct}}\right)=$ $0.354 \mathrm{~mm}^{-1} ; R=7.95 \%$. The molecule is shown to be the pentacyclic triterpene $9(11)$-fernene, consisting of

^[ *IUPAC name: ent-13.17-dimethyl-26,28-dinor-5 $\beta$.10a hop-9(11)-ene. ]


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four fused all-trans six-membered rings $A, B, C, D$ with a five-membered $E$ ring trans-fused to ring $D$. Ring $B$ is intermediate between twist and boat conformations.

Introduction. In the course of the investigation of the unknown allergenic compound of the leatherleaf fern Arachnoides adiantiformis (Forst) Tindale (fam. Aspidiaceae) (Hausen \& Schulz, 1978) the title
compound has been found among the allergenic fraction as the only acceptable crystal for X-ray structure analysis. The identification by X-ray analysis as 9(11)-fernene has not previously been reported. As a triterpenoid hydrocarbon of a new type (Ageta, Iwata \& Natori, 1963) this fernene has been identified by chemical methods in several fern species (Horveth, de Szöcs, Alvarado \& Grant, 1975; Khosa, Wahi \& Mukherjee, 1978; Wij, 1978; Das Gupta \& Khastgir, 1978; Khastgir \& Shoolery, 1978), in a moss (Catalano, Marsili, Morelli \& Pacchiani, 1976) and even in a yew tree (Silva, Hoeneisen \& Sammes, 1972). A small colourless crystal $0.15 \times 0.15 \times 0.08$ mm was mounted along a for the measurements. The space group and symmetry were determined by photography, and cell parameters obtained by leastsquares refinement from the angular settings of 12 reflections on a Syntex $P 2_{1}$ four-circle computercontrolled diffractometer with graphite-monochromated Cu Kor radiation. The intensities for 2107 independent reflections with $2 \theta \leq 115^{\circ}$ were measured on the diffractometer by the $\theta-2 \theta$ scan technique at a variable scan rate of 1 to $29^{\circ} \min ^{-1} .1383$ of these reflections were observed above background with $I>$ $2 \sigma$. Data reduction (with Lp corrections) was carried out with $L S A W$ (Eck, 1975).

The calculation of normalized structure factors and the solution of the structure were performed with MULTAN 78 (Main, Hull, Lessinger, Germain, Declercq \& Woolfson, 1978). The density of the initially unknown specimen was estimated to be 1.25 $\mathrm{Mg} \mathrm{m}{ }^{-3}$ with $M_{r}=496$ for $Z=4 \mathrm{C}_{24} \mathrm{H}_{36} \mathrm{O}_{11}$. The $E$ map based on $392 E$ 's $\geq 1.24$ with phases in the most consistent set (combined figure of merit $=2.97$ ) showed 32 independent maxima. After refinement of the positional and isotropic temperature parameters and calculation of difference syntheses with SHELX (Sheldrick, 1975), five peaks had to be rejected and three other non-hydrogen atomic positions were located. Four six-membered rings connected with a five-membered ring could be established ( $R=18.4 \%$ ). Bond lengths and angles, and temperature factors as well as site-occupation factors led to the conclusion that the structure must be a hopane-derivative triterpenoid. Furthermore, the 9(11) double bond of $\sim 1.34 \AA$ and the conformation of the methyl groups unequivocally established this compound to be 9(11)fernene. All the 50 expected H atoms except H (162), $\mathrm{H}(283)$ and $\mathrm{H}(301)$ were found after refinement of the positional and anisotropic temperature parameters of the C atoms ( $R=14.4 \%$ ) and calculation of three more difference maps. In the final full-matrix leastsquares refinement, when the separation of variables into smaller groups was necessary, positional parameters for all atoms and anisotropic temperature parameters for the C atoms were varied. The isotropic temperature factors for all H atoms were fixed at $U=$

Table 1. Final positional parameters $\left(\times 10^{4}\right)$ with e.s.d.'s in parentheses

|  | $x$ | $y$ | $z$ |
| :--- | :--- | :---: | :---: |
|  |  |  |  |
| $\mathrm{C}(1)$ | $7195(11)$ | $1658(12)$ | $3353(3)$ |
| $\mathrm{C}(2)$ | $6889(16)$ | $2042(17)$ | $2891(4)$ |
| $\mathrm{C}(3)$ | $5104(15)$ | $2737(12)$ | $2858(3)$ |
| $\mathrm{C}(4)$ | $3611(1)$ | $1848(7)$ | $2995(2)$ |
| $\mathrm{C}(5)$ | $4024(9)$ | $1314(7)$ | $3436(2)$ |
| $\mathrm{C}(6)$ | $2528(10)$ | $516(9)$ | $3615(3)$ |
| $\mathrm{C}(7)$ | $3032(11)$ | $-222(9)$ | $4007(3)$ |
| $\mathrm{C}(8)$ | $4548(8)$ | $307(6)$ | $4248(2)$ |
| $\mathrm{C}(9)$ | $6135(9)$ | $473(7)$ | $3980(2)$ |
| $\mathrm{C}(10)$ | $5824(9)$ | $705(7)$ | $3510(2)$ |
| $\mathrm{C}(11)$ | $7750(8)$ | $330(8)$ | $4134(2)$ |
| $\mathrm{C}(12)$ | $8144(8)$ | $97(7)$ | $4592(2)$ |
| $\mathrm{C}(13)$ | $6576(8)$ | $333(6)$ | $4871(2)$ |
| $\mathrm{C}(14)$ | $4935(8)$ | $-331(6)$ | $4672(2)$ |
| $\mathrm{C}(15)$ | $3406(8)$ | $-284(6)$ | $4975(2)$ |
| $\mathrm{C}(16)$ | $3776(9)$ | $-809(6)$ | $5420(2)$ |
| $\mathrm{C}(17)$ | $5346(8)$ | $-199(6)$ | $5627(2)$ |
| $\mathrm{C}(18)$ | $6880(8)$ | $-195(6)$ | $5317(2)$ |
| $\mathrm{C}(19)$ | $8443(9)$ | $202(8)$ | $5568(2)$ |
| $\mathrm{C}(20)$ | $8110(9)$ | $-400(8)$ | $6005(2)$ |
| $\mathrm{C}(21)$ | $6233(11)$ | $-921(6)$ | $5994(2)$ |
| $\mathrm{C}(22)$ | $5367(12)$ | $-941(8)$ | $6430(3)$ |
| $\mathrm{C}(23)$ | $3273(16)$ | $919(10)$ | $2655(3)$ |
| $\mathrm{C}(24)$ | $1970(13)$ | $2644(9)$ | $3019(3)$ |
| $\mathrm{C}(25)$ | $6105(16)$ | $-505(11)$ | $3275(3)$ |
| $\mathrm{C}(26)$ | $6275(9)$ | $1791(6)$ | $4886(2)$ |
| $\mathrm{C}(27)$ | $5299(9)$ | $-1744(6)$ | $4582(2)$ |
| $\mathrm{C}(28)$ | $4820(9)$ | $1141(7)$ | $5791(2)$ |
| $\mathrm{C}(29)$ | $3502(15)$ | $-1470(11)$ | $6420(3)$ |
| $\mathrm{C}(30)$ | $6418(15)$ | $-1691(10)$ | $6736(3)$ |

$0.07 \AA^{2}$. The final value of $R$ was $7.95 \%$, based on unit weights ( $R=\sum| | F_{o}\left|-\left|F_{\mathrm{c}}\right|\right| / \sum\left|F_{o}\right|$ ). The positional parameters of the C atoms are given in Table 1.* A final difference synthesis resulted in peak heights $<0.58$ e $\AA^{-3}$ without any chemical significance.

Discussion. The structure is shown in Fig. 1. The bond distances, valency angles and endocyclic torsion angles involving the C atoms only are given in Figs. 2-4. The largest e.s.d. for the bond lengths is $0.019 \AA$, for the valency angles $1.2^{\circ}$ and for the torsion angles $2 \cdot 2^{\circ}$. The steric arrangement of the methyl groups and H atoms (Fig. 5) is also found in retigerate A (Takahashi \& Iitaka, 1972), davallic acid (Nakanishi, Lin, Kakisawa, Hsü \& Hsiu, 1963) and partly in arundoin (Nishimoto, Ito, Natori \& Ohmoto, 1965). The chair-shaped rings $A$ and $D$ are distorted; Fig. 4 shows the deviation of the torsion angles from the standard value of $55.9^{\circ}$ (Geise, Buys \& Mijlhoff, 1971). Nakanishi et al. (1963) compared fernene with davallic

[^1]

Fig. 1. ORTEP (Johnson, 1971) stereodrawing of the title compound.


Fig. 2. Bond distances $(\AA) ; \sigma$ range $=0.009-0.019 \AA$.


Fig. 3. Bond angles $\left({ }^{\circ}\right) ; \sigma$ range $=0 \cdot 5-1 \cdot 2^{\circ}$.
acid and postulated with the aid of Dreiding models that ring $B$ was fixed in a boat form. From this structure determination ring $B$ is intermediate between boat and twist conformations, while the cyclohexene ring $C$ has a slightly distorted half-chair form. Ring $E$ shows a slightly distorted envelope conformation with


Fig. 4. Endocyclic torsion angles $\left({ }^{\circ}\right)$. A torsion angle $a-b-c-d$ is positive (viewed down the $b-c$ bond) if the $a-b$ bond will eclipse the $c-d$ bond by a rotation of less than $180^{\circ}$ in a clockwise direction.


Fig. 5. Perspective view approximately along the mean molecular plane.
torsion angles similar to those in hortensyl $p$ iodobenzoate (McCandlish \& Stout, 1976). The mean deviation of $\mathrm{C}(18)$ to $\mathrm{C}(21)$ from the best plane is $0.045 \AA ; \mathrm{C}(17)$ deviates $0.694 \AA$ from this plane, calculated with XANADU (Roberts \& Sheldrick, 1975).

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# Structure of (2RS,5RS,8RS,11SR)-1,4,7,10-Tetrabenzyl-2,5,8,11-tetraethyl-1,4,7,10tetraazacy clododecane 

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#### Abstract

C}_{44} \mathrm{H}_{60} \mathrm{~N}_{4}\), triclinic, $P \overline{1}, a=11.56$ (1), $b=$ 18.60 (2), $c=9.754$ (5) $\AA, a=105.10$ (6),,$\beta=$ 106.68 (4), $\gamma=85.00$ (5) ${ }^{\circ}, U=1939.0$ (2) $\AA^{3}, Z=2$, $D_{m}=1 \cdot 10, D_{c}=1.104$ (1) $\mathrm{Mg} \mathrm{m}^{-3}$. The crystal structure was solved by the Monte Carlo direct method. The 12 -membered ring has a distorted square conformation with the methylene groups at the corners.


Introduction. This paper is part of a series of investigations on the cyclic tetramers of chiral aziridines. These tetramers can be divided into four geometrical isomers, viz $C_{4}, C_{1}, S_{4}$ and $C_{i}$ (Tsuboyama, Tsuboyama, Uzawa \& Higashi, 1974). Among these, the crystal structures of $C_{4}(R R R R$-type $)$ (Sakurai, Kobayashi, Tsuboyama \& Tsuboyama, 1978a,b), and $C_{i}$ (RRSS-type) (Hiramatsu, Sakurai, Tsuboyama \& Tsuboyama, 1979) have been reported. In this paper the structure of the title compound with $C_{1}$ symmetry ( $R R R S$-type) is described. Triclinic single crystals of the racemic compound ( $R R R S$ and SSSR) were obtained from ethyl acetate solution. A clear colourless plate crystal with dimensions $0.4 \times 0.4 \times$ 0.2 mm was used for the study. X-ray diffraction data were measured on a Rigaku AFC four-circle diffractometer, with graphite-monochromatized Mo Ka radiation. Within the range $2 \theta \leq 50^{\circ}, 4583$ independent reflections with $|F|>3 \sigma(F)$ were obtained. The intensities were corrected for Lorentz and polarization factors, but no absorption correction was applied.

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The first attempt to solve the structure by MULTAN (Main, Woolfson \& Germain, 1971) was not successful. The Monte Carlo method (Furusaki, 1979) was then applied using 24 initial sets, and a reasonable solution was obtained through the FASTAN part of MULTAN.

The structure was refined by the block-diagonal least-squares method. Unit weight was given to all reflections, and anisotropic temperature factors were used for all non-hydrogen atoms. All H atoms were located by a difference Fourier synthesis, and included in the final refinement with isotropic temperature factors. The final $R$ index was $5.4 \%$. Atomic parameters are given in Table 1.*


Fig. 1. Stereoscopic drawing of the molecule projected along the normal to the 12 -membered ring.
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[^1]:    * Lists of structure factors, anisotropic thermal parameters of the C atoms and positional parameters of the H atoms have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 35248 (11 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography. 5 Abbey Square. Chester CHI 2HU, England.

